

FE0007632: Novel Inorganic/Polymer Composite Membranes for CO₂ Capture

PI: Winston Ho, Professor

**Department of Chemical & Biomolecular Engineering
Department of Materials Science and Engineering**

Co-PI: Prabir Dutta, Professor

**Department of Chemistry
The Ohio State University**

**Subcontractor: Steve Schmit, Director of R&D
Gradient Technology**

**2013 NETL CO₂ Capture Technology Meeting
Pittsburgh, PA, July 8 – 11, 2013**

Funding and Performance Dates

- **Total Budget: 10/01/2011 – 05/31/2015**
DOE: \$3,000K; OSU: \$679K; ODOD: \$500K
- **BP1: 10/01/2011 – 05/31/2013**
DOE: \$899K; OSU: \$351K
- **BP2: 06/01/2013 – 05/31/2014**
DOE: \$958K; OSU: \$131K; ODOD: \$277K
- **BP3: 06/01/2014 – 05/31/2015**
DOE: \$1,144K; OSU: \$197K; ODOD: \$223K

Project Organization and Roles

Ohio State University

- Technical lead
- Concept development and execution
- Novel membrane synthesis/characterization
- Membrane scale-up/continuous fabrication
- Process demonstration
- Cost calculations

Winston Ho and Prabir Dutta

DOE NETL

Project Manager

José Figueroa

Gradient Technology

- System, cost analysis
- EH&S analysis

Steve Schmit

TriSep Corporation

- Consult on continuous membrane fabrication

Peter Knappe

AEP

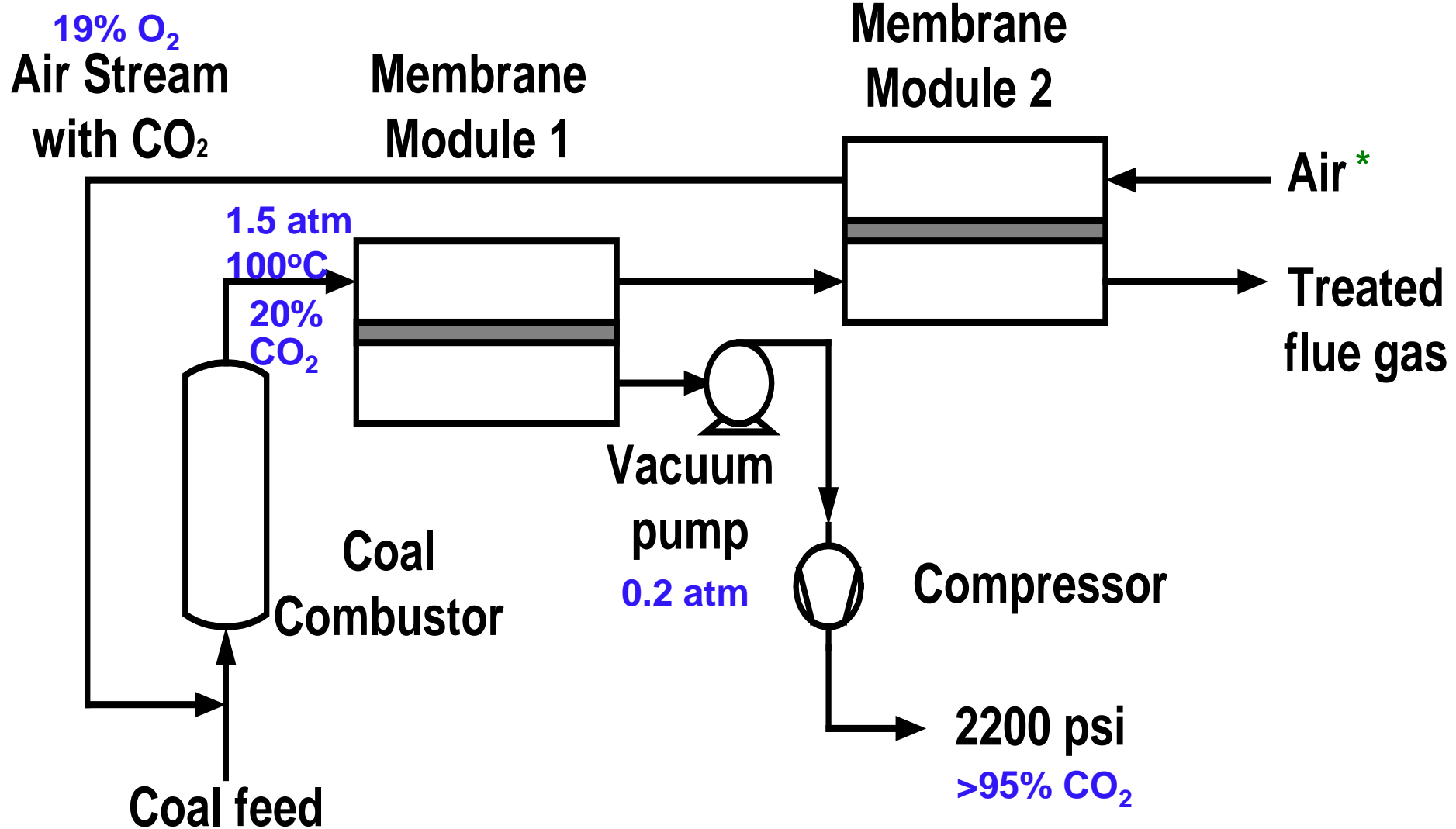
- Consult on plant integration, demonstration and EH&S

Dan Duellman

Project Objective

- **Develop cost-effective design and manufacturing process for new membrane modules that capture CO₂ from flue gas with <35% COE increase and <\$40/tonne CO₂**
- **BP1**
 - Bench scale membrane synthesis, characterization, downselection, and gas separation performance
 - Preliminary techno-economic analysis
- **BP2**
 - Bench scale membrane synthesis, characterization and gas separation performance to continue
 - Continuous membrane fabrication
 - Membrane module testing in lab (CO₂, N₂, MOISTURE)
 - Update techno-economic analysis
- **BP3**
 - 3 prototype modules for testing with simulated flue gas
 - Update techno-economic analysis
 - EH&S evaluation report will be developed

Process Proposed for CO₂ Capture from Flue Gas in Coal-Fired Power Plants



*Air Sweep first used by MTR

Membrane Approaches

- **Two Composite Membrane Approaches**
 - All Inorganic composite membrane
 - Zeolite-Y/Polymer composite membrane
 - + Amino based cover layer
 - + Non-Amino based cover layer

Current Project Success Criteria

- **BP1**

- For polymer support:
1000 GPU Permeance with Selectivity = 50
- For ceramic support:
1000 GPU Permeance with Selectivity = 100

Current state-of-the-art 2000 GPU, selectivity = 50

- **BP2**

- For polymer support:
1500 – 2000 GPU Permeance with Selectivity = 50 – 75
- SO₂ effects on membrane (1 – 50 ppm), determine stability of amine membrane with time (100 hours), define limits (24 hours) if polishing unit fails

- **BP3**

- For polymer support:
2500 – 3000 GPU Permeance with Selectivity = 100 – 150

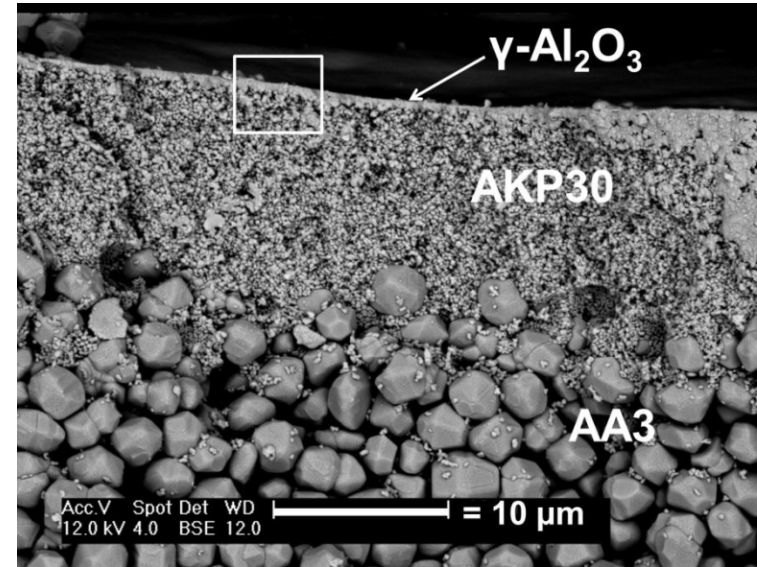
Budget Period 1

- **Two Types of Inorganic Selective Layers Investigated**
 - Alumina
 - Zeolite-Y

- **Alumina Selective Layer on Two Supports**
 - Alumina Selective Layer on Ceramic Support
 - Alumina Selective Layer on Polymer Support

All Inorganic Composite Membrane

- Highly permeable ceramic support prepared
- Rapid modification method of γ -alumina did not give adequate performance (separation factors <10)
- Low selectivity due to
 - Defects, Cracks
- Difficult to scale-up
 - Inorganic support
- Continuous fabrication
 - Not suitable
 - High cost

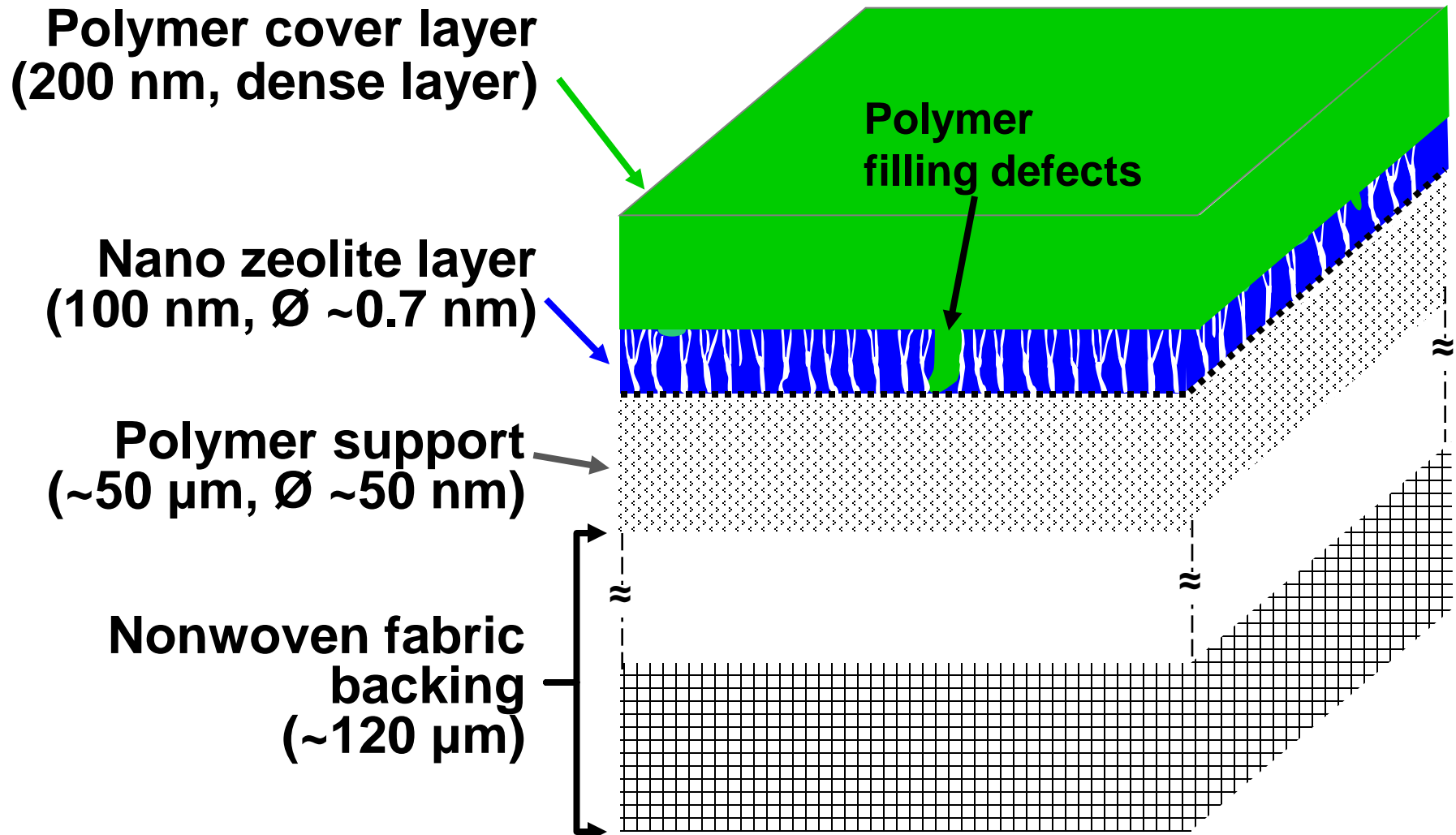


- Overall results led to down selection (October 2012) of Zeolite-Y/Polymer composite membrane

Zeolite-Y/Polymer Composite Membranes

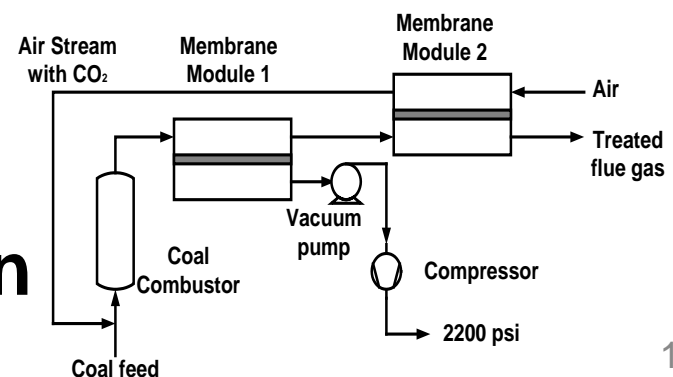
- **Zeolite-Y has good affinity to CO₂**
- **Two Polymer Strategies**
 - **Amine cover layer**
 - **Non-amine cover layer**
- **Polymer support**
 - **Suitable for continuous fabrication**
 - **Low processing cost**

Inorganic/Polymer Composite Membrane to Capture High Inorganic Performance and Low-Cost Polymer Processing

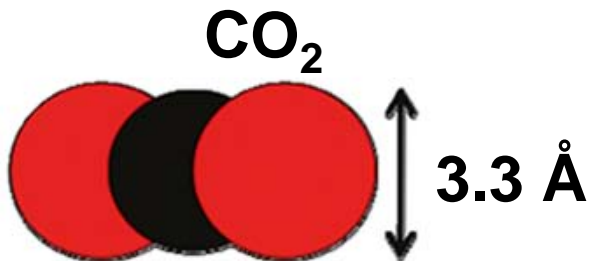
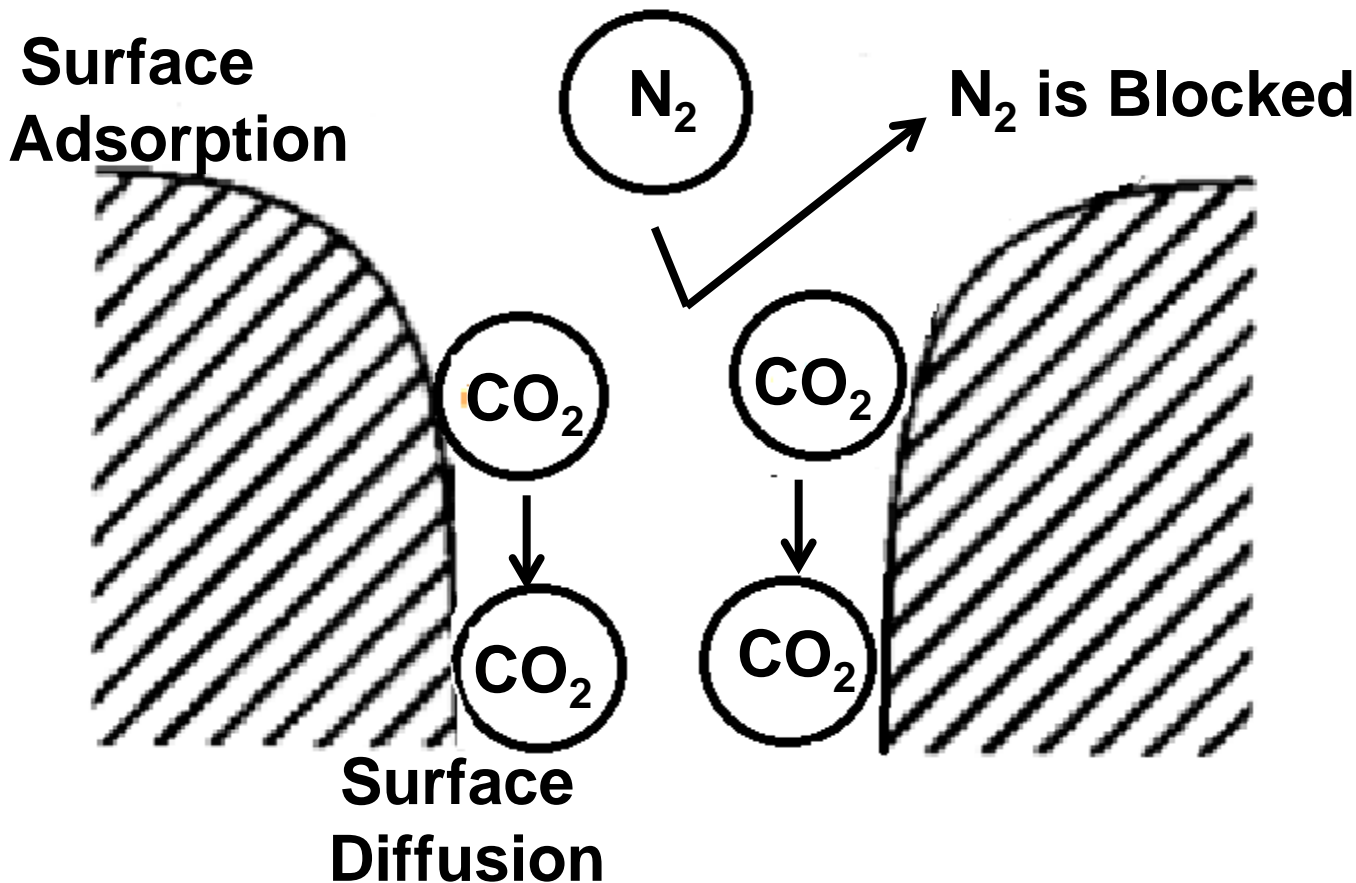


BP1 Accomplishments

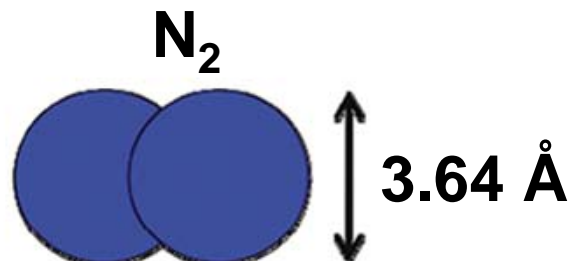
- **Zeolite/amine polymer composite membranes showed 1100 GPU with ~800 selectivity**
 - Achieved BP1 target of 1000 GPU & 50 selectivity
- **Significant Membrane Synthesis Improvements**
 - Including discovery of rapid zeolite synthesis (< 1 hour) for continuous membrane fabrication
 - Ceramic/polymer spiral-wound membrane element exhibited at 2012 NETL meeting; Zeolite/polymer element also rolled successfully
 - Ceramic membrane processing also improved (20 min in lieu of 43 hours), but membrane performance poor
 - Two polymer approaches, amine and non-amine membranes demonstrated
- **Preliminary techno-economic calculations (1100 GPU and >200 selectivity) showed ~46.0% COE increase and ~\$27.7/tonne CO₂**
 - Techno-economic model published
- **Proposed membrane process eliminates cryogenic distillation (compare to competition)**



Transport Mechanism through Zeolite



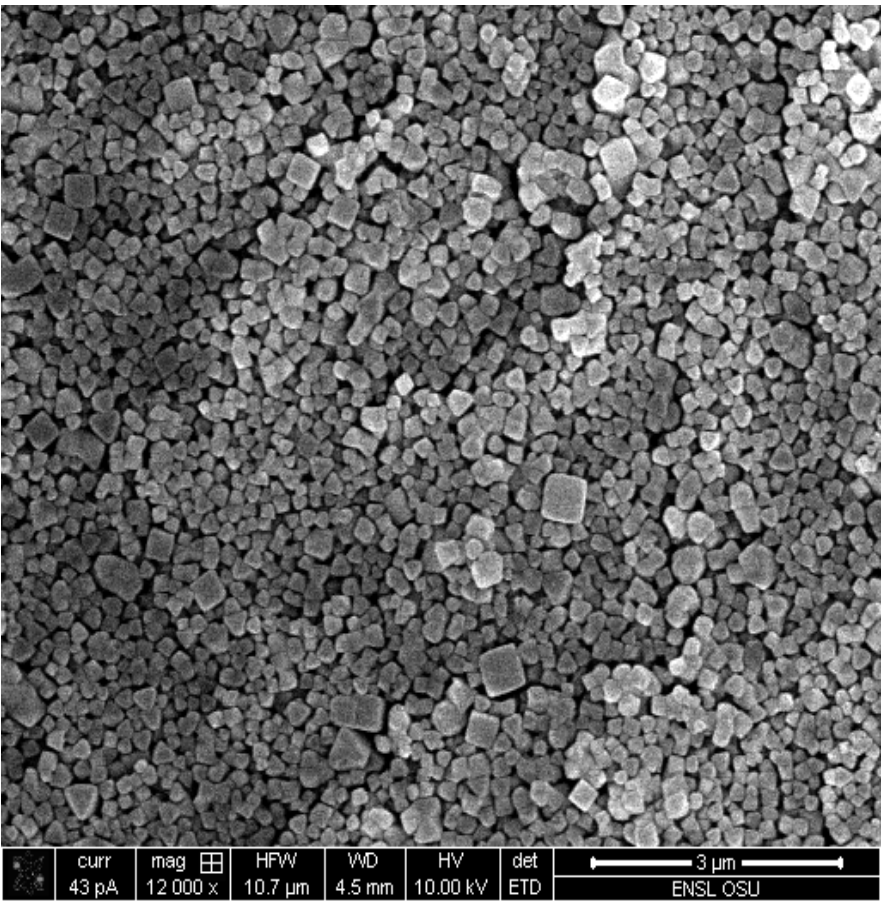
$$Q_{\text{CO}_2} = -13.67 \times 10^{-40} \text{ C m}^2$$



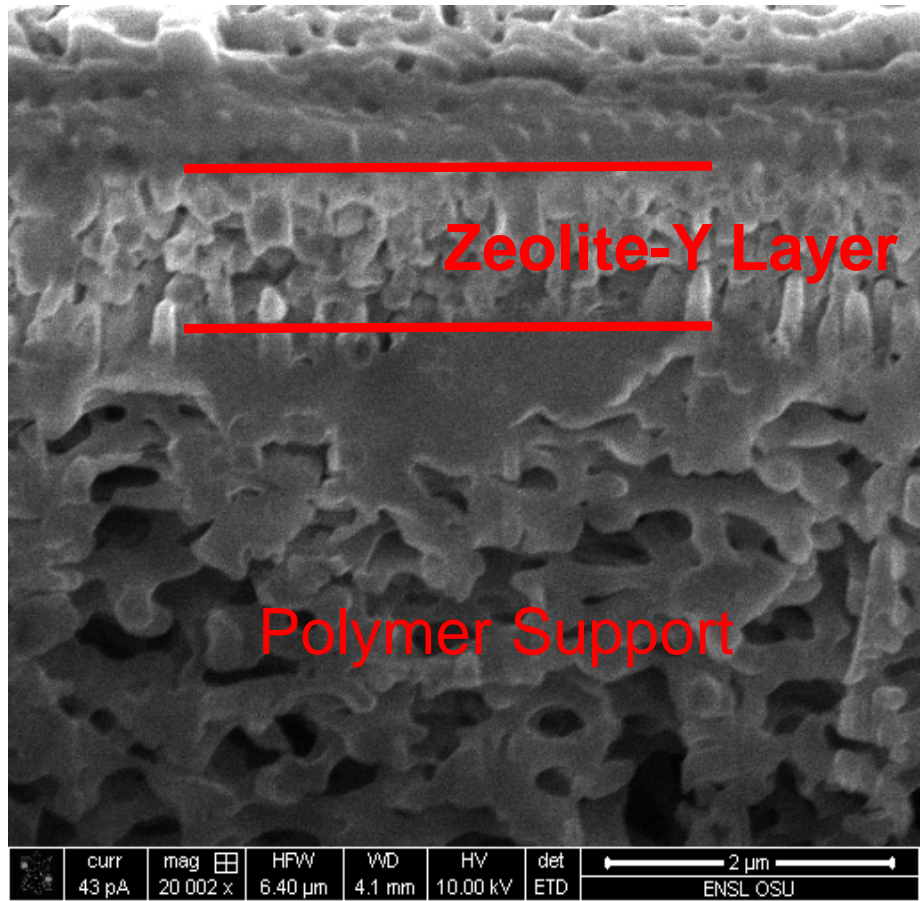
$$Q_{\text{N}_2} = -4.67 \times 10^{-40} \text{ C m}^2$$

Zeolite Nanoparticles Deposited on Polymer Support Successfully

Top View



Cross-section

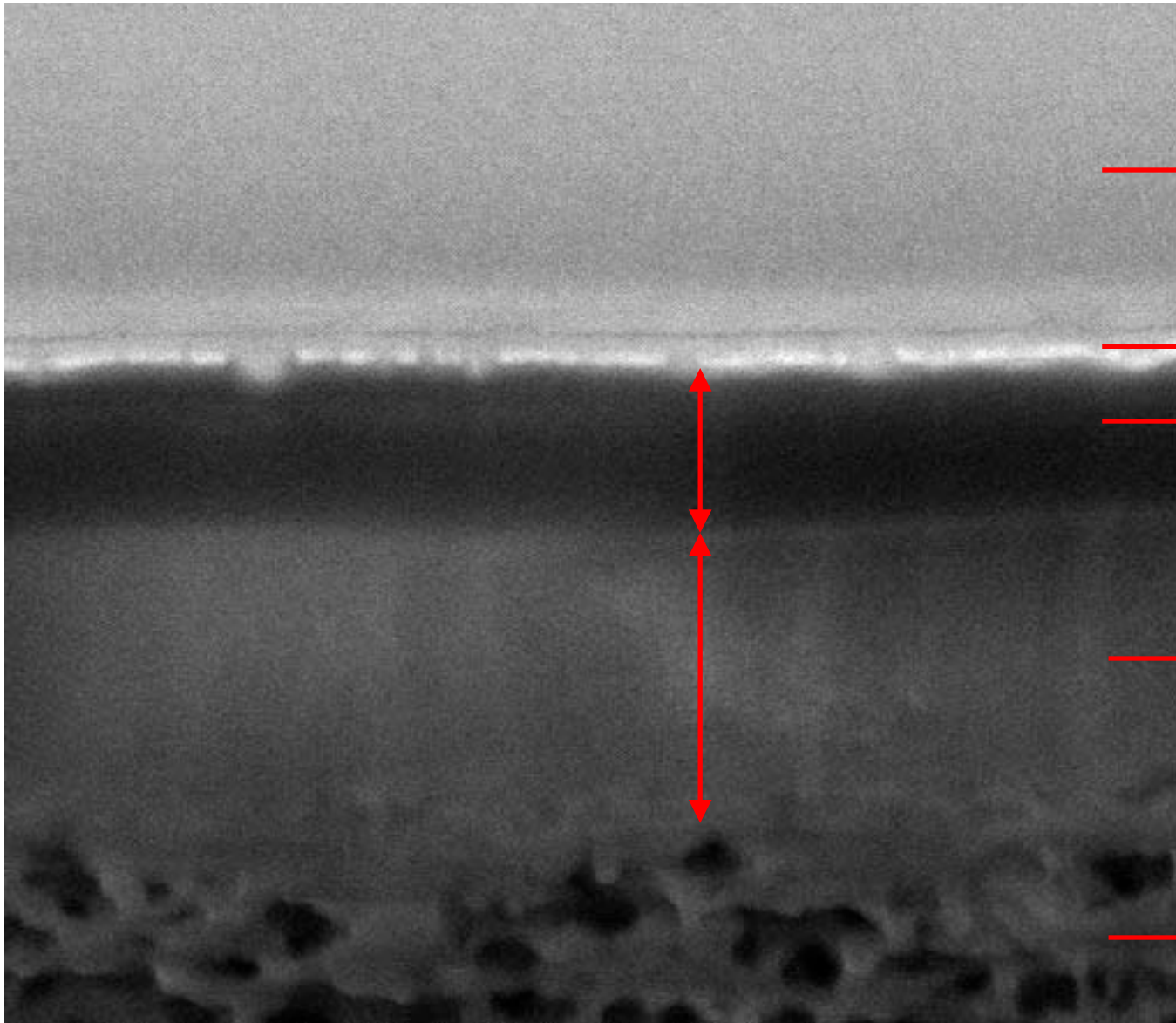


Two Cover Layer Approaches for Zeolite/ Polymer Composite Membranes

- **Amine cover layer**
- **Non-amine cover layer**

Amine/Zeolite Seed Layer/Polymer Support

Platinum and gold used to provide conductivity for focused ion beam microscopy



Pt deposition for better conductivity

Gold coating

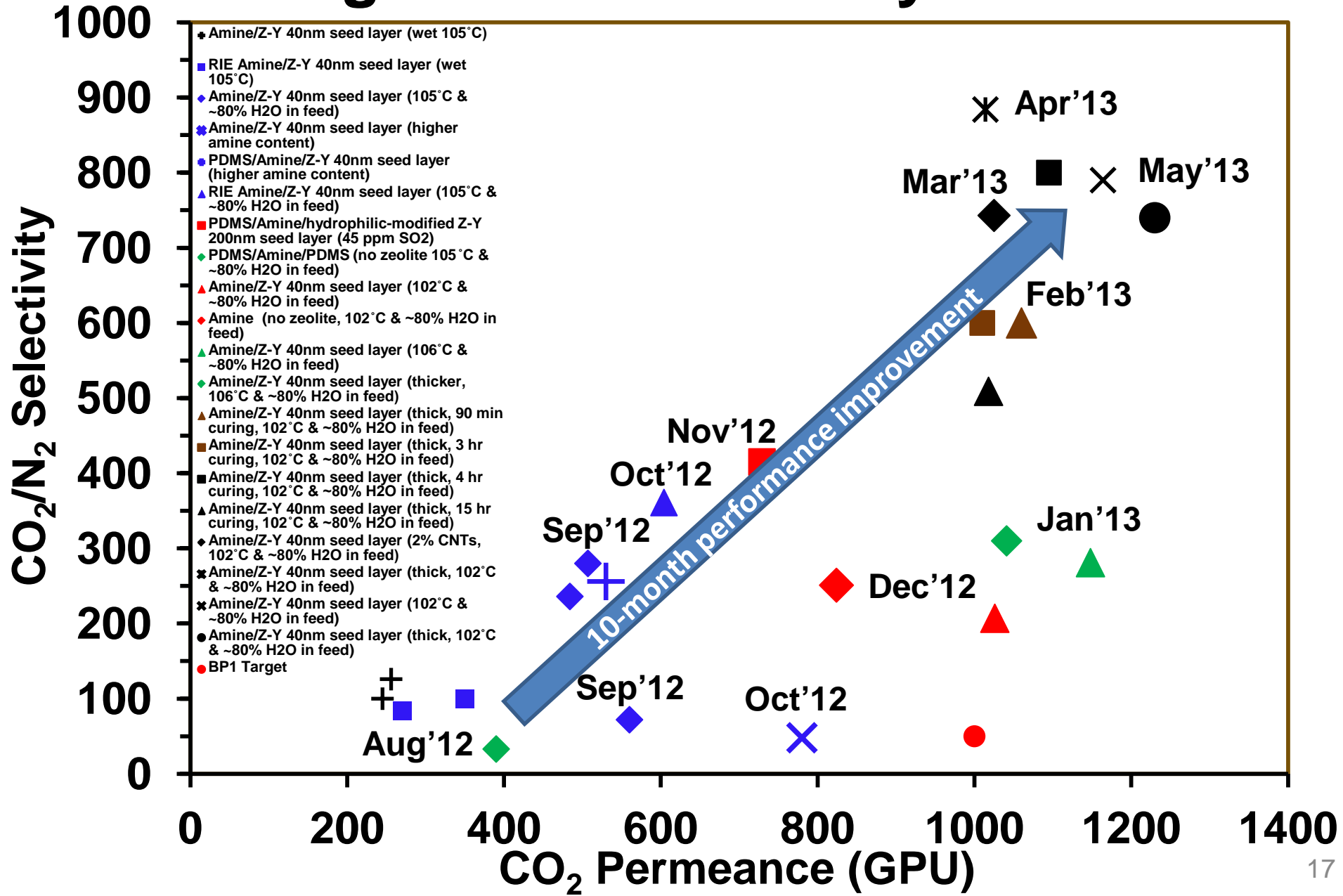
Amine cover layer
~ 500 nm

Zeolite-Y 40 nm seed layer
< 1 μm

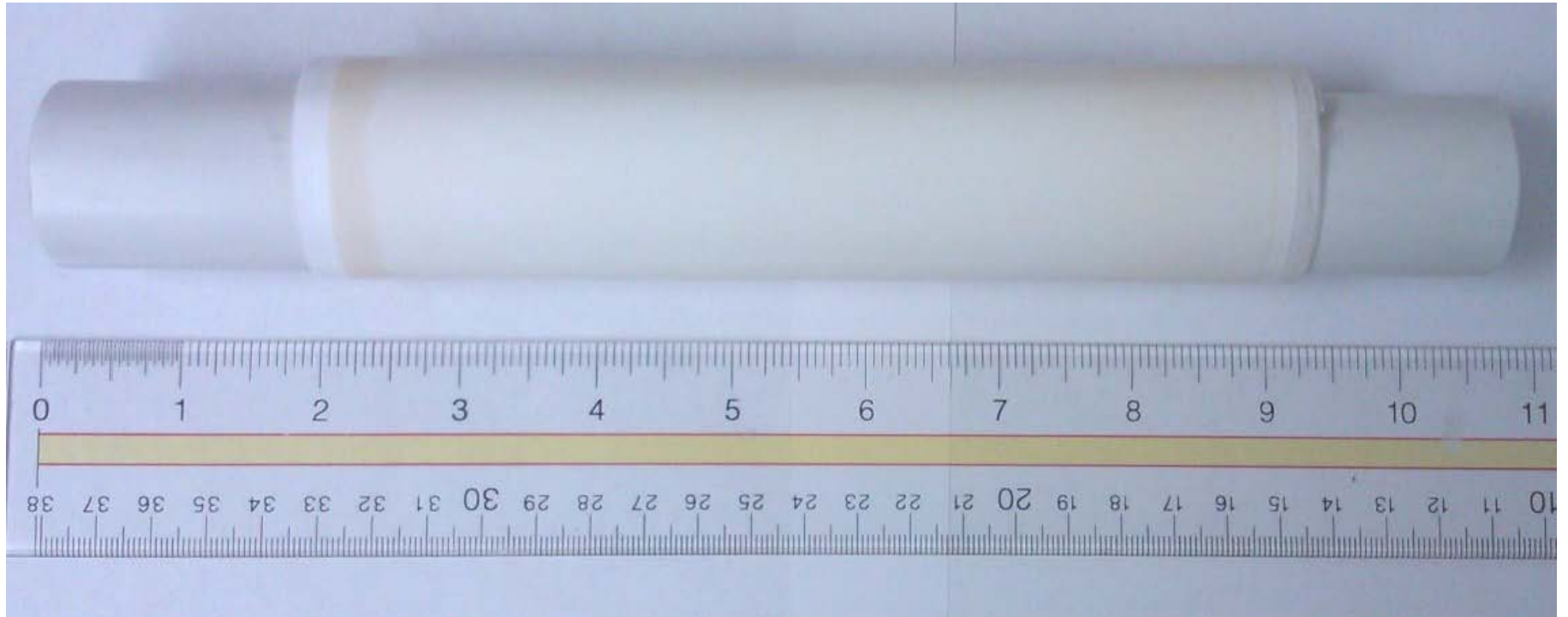
PES support

	curr 86 pA	mag 35 007 x	HPFW 3.66 μm	WD 4.2 mm	HV 5.00 kV	det ETD	 1 μm ENSL OSU
--	---------------	-----------------	-----------------	--------------	---------------	------------	--

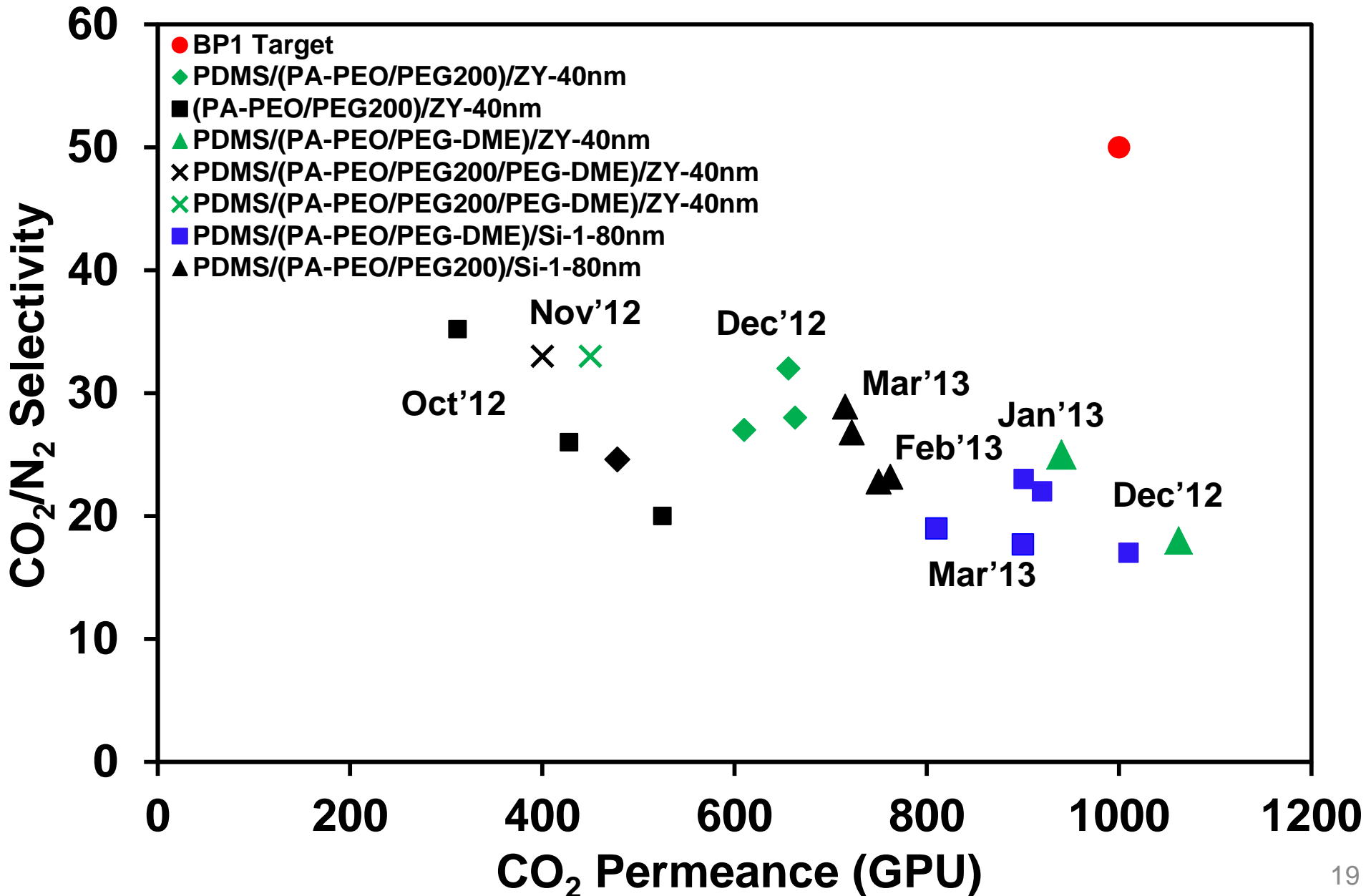
Zeolite-Polymer Composite Membranes Containing Amine Cover Layer at ~102°C



Zeolite/Polymer Spiral-Wound Membrane Element Rolled Successfully



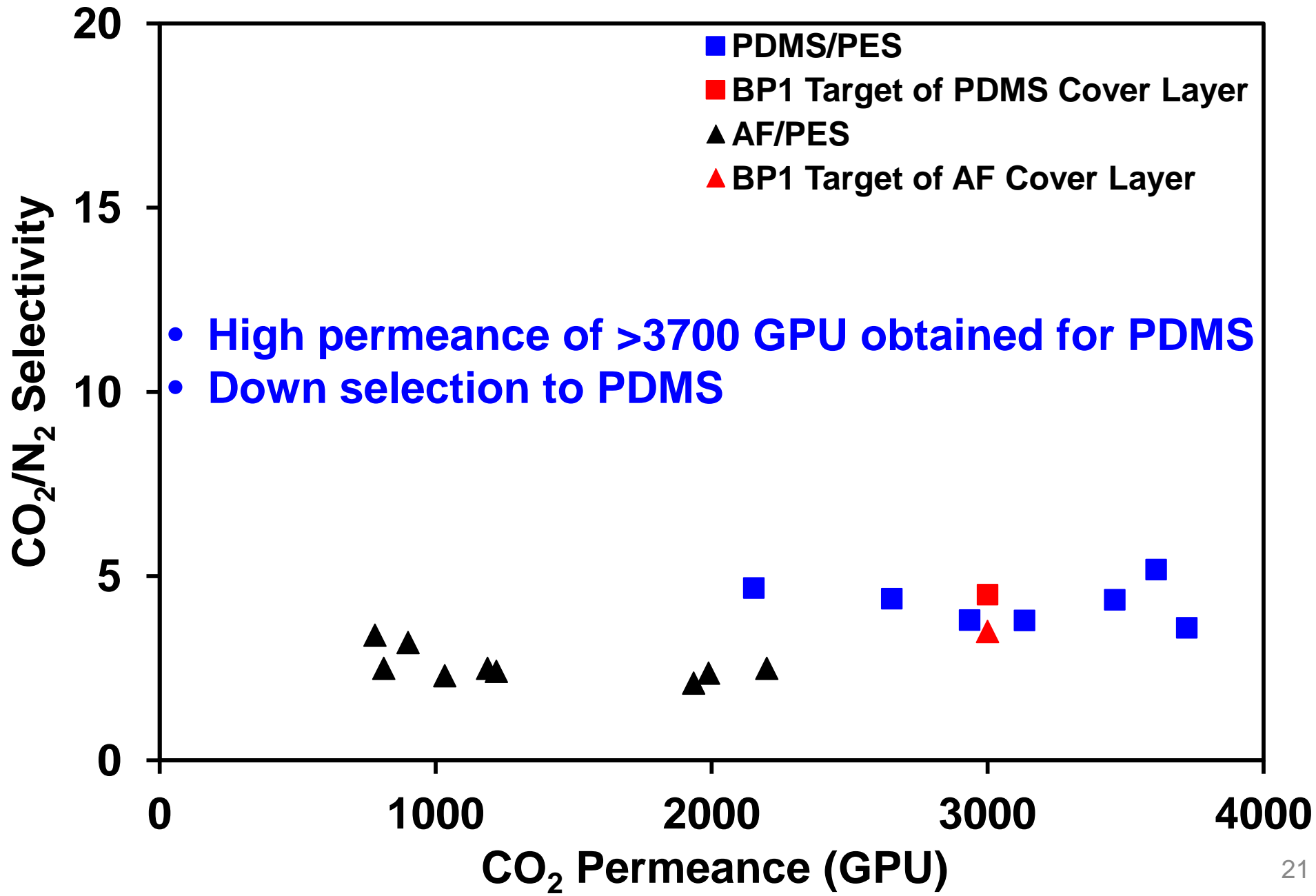
Zeolite-Polymer Composite Membranes with SO₂-Insensitve Cover Layer at 57°C



Additional Polymer Cover Layer Approaches for Defect Abatement

- **PDMS (polydimethylsiloxane) cover layer**
- **AF (amorphous fluorinated) polymer cover layer**

Polymer Cover Layers for Defect Abatement



Cost Calculations

Assumptions

- **CO₂ Permeance: 1100 GPU, CO₂/N₂ Selectivity >200**
 - Based on present membrane lab data with amine membrane
- **Operating Feed Pressure of 1.5 atm at ~100°C**
 - Permeate pressure of 0.2 atm for Stage 1 and 1 atm for Stage 2 (air sweep)
 - Selectivity of >200 under these conditions has negligible effect on costs
 - Pressure recovery using turboexpander

Results

- **Increase on Cost of Electricity (COE) = 46.0% with 1100 GPU**
Capture Cost = \$27.7/tonne CO₂
- **If CO₂ Permeance = 3000 GPU**
 - COE increase is 38.2% with a capture cost of \$23.7/tonne CO₂

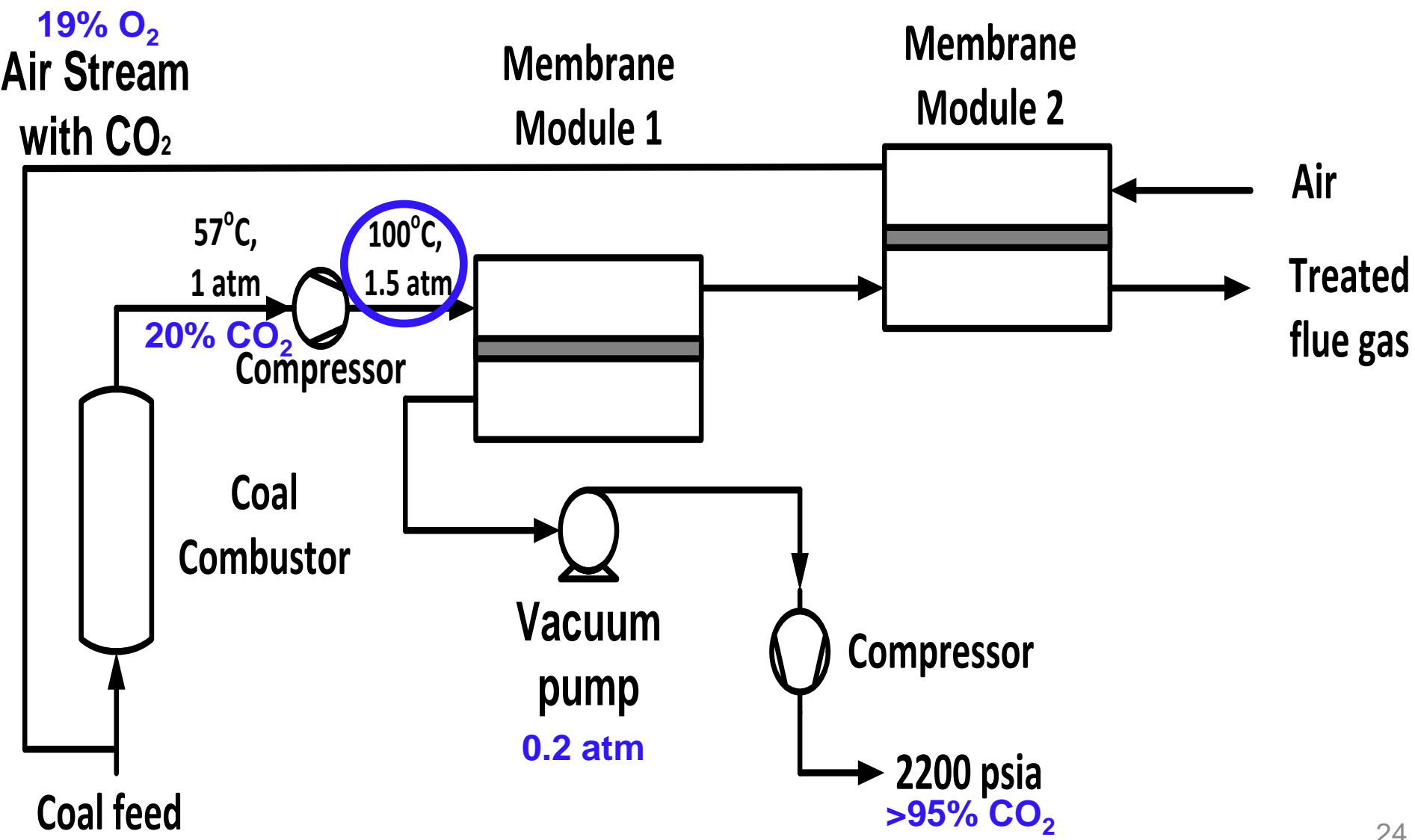
State-of-the-Art Amine Scrubbing

- **86% Increase on COE and Capture Cost of \$71/ton CO₂**

BP1 Technical Research Needs

- **Temperature Effect**
 - Amine Membrane operational at 102°C
 - Compression from 1 atm to 1.5 atm needed for cost optimization
 - Temperature increases from 57°C to 100°C
- **SO₂ Effects and Solutions**
 - Approach 1: SO₂-Sensitive Amine Polymer Cover Layer meets BP1 Goals
 - SO₂ Polishing Step required
 - Approach 2: SO₂-Insensitive Polymer Cover Layer meets BP1 Permeance Goal
 - But selectivity (30) needs to be improved

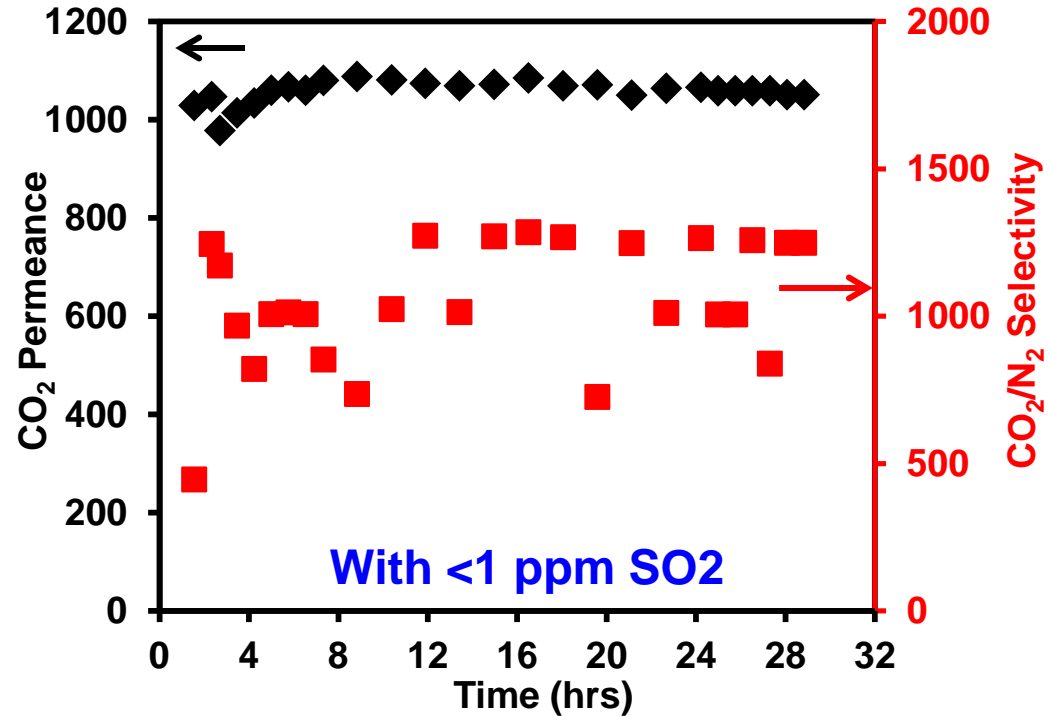
Temperature Mitigation for CO₂ Capture from Flue Gas in Coal-Fired Power Plants



SO₂ Effects on Amine-containing Membranes

• SO₂ Effects

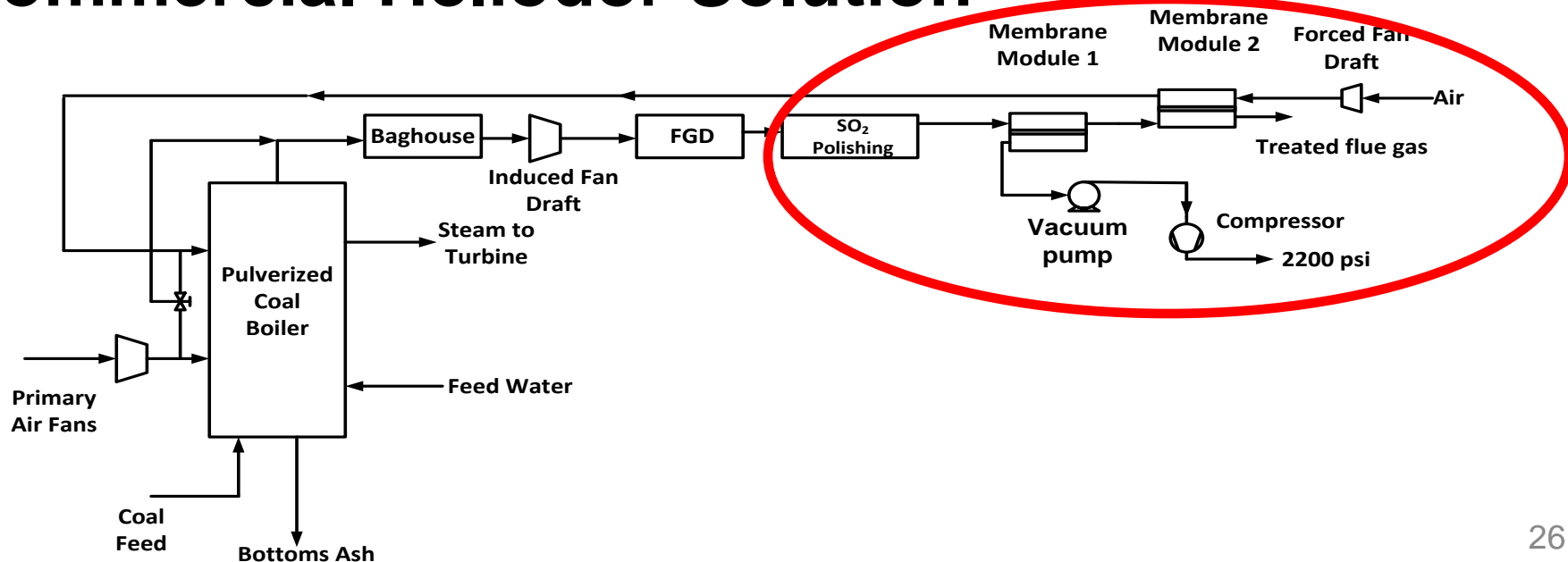
- SO₂ at ≥ 10 ppm appeared to affect stability of membrane containing amine cover layer
- SO₂ at ≤ 1 ppm appeared not to affect stability of membrane with amine cover layer
- More study required between 1 – 10 ppm SO₂



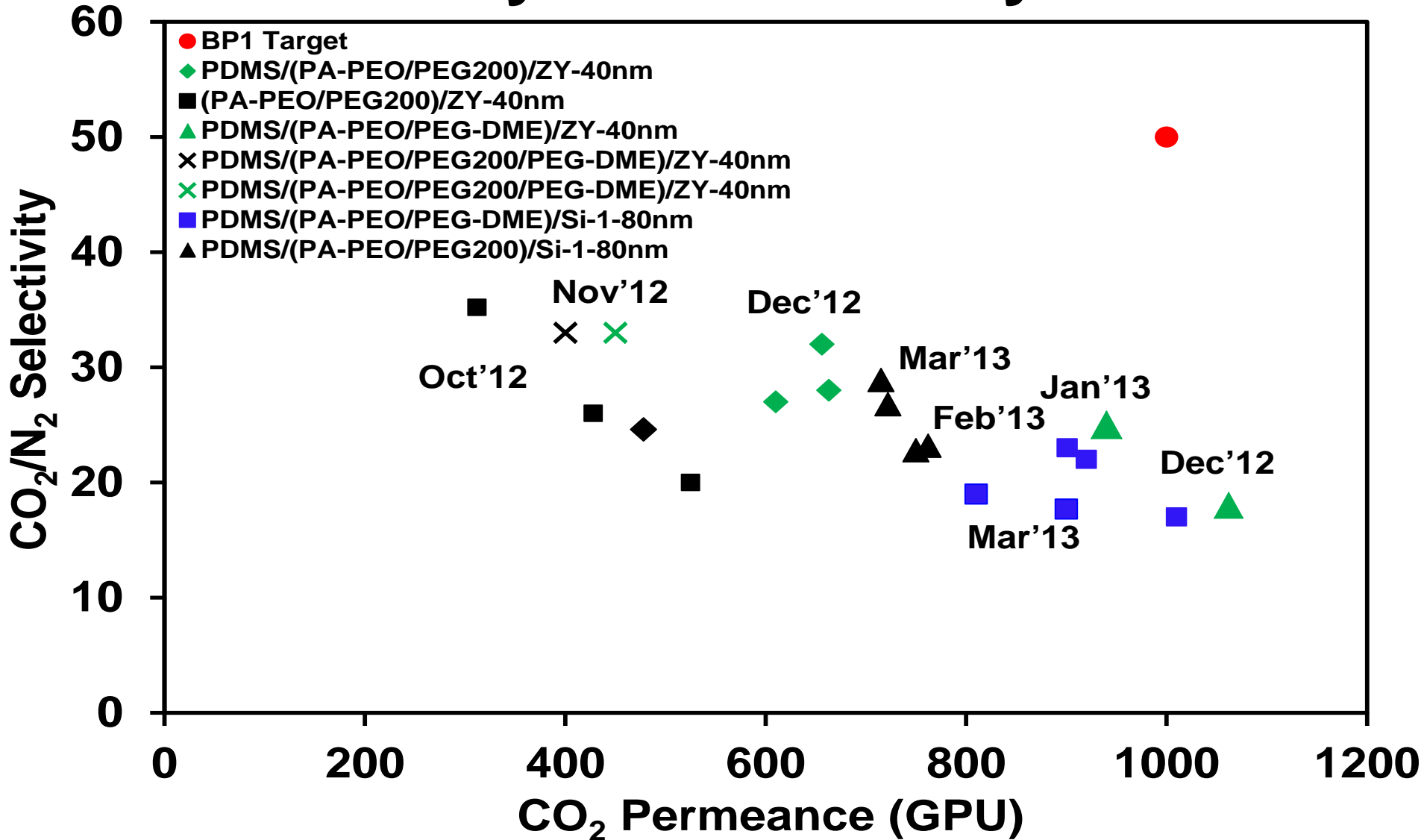
- Propose SO₂ Polishing Step before membrane

Approach 1: SO₂ Membrane Mitigation raised in BP 1

- **Absorption into 20 wt% NaOH Solution**
 - + Polishing step based on NETL baseline document
 - ++ Estimated to be about <1.5% increase on COE
 - + Non-plugging, low-differential-pressure, spray baffle scrubber
 - + Efficiencies can reach >95%
 - + SO₂ can be reduced from 44 ppm to <2 ppm
- **Commercial ZnO for SO₂ Clean-up to <0.1 ppm**
- **Commercial Heliodor Solution**



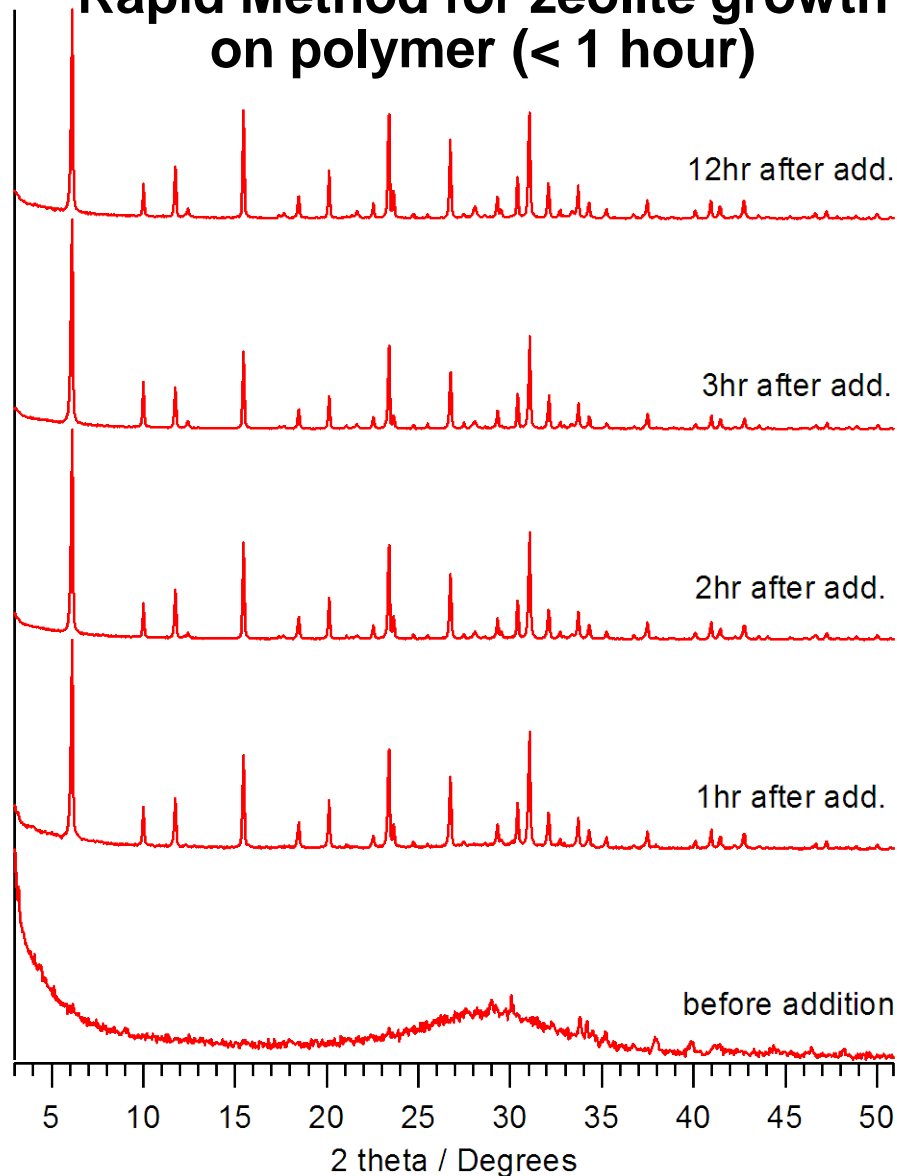
Approach 2: Develop SO₂-Insensitive Polymer Cover Layer



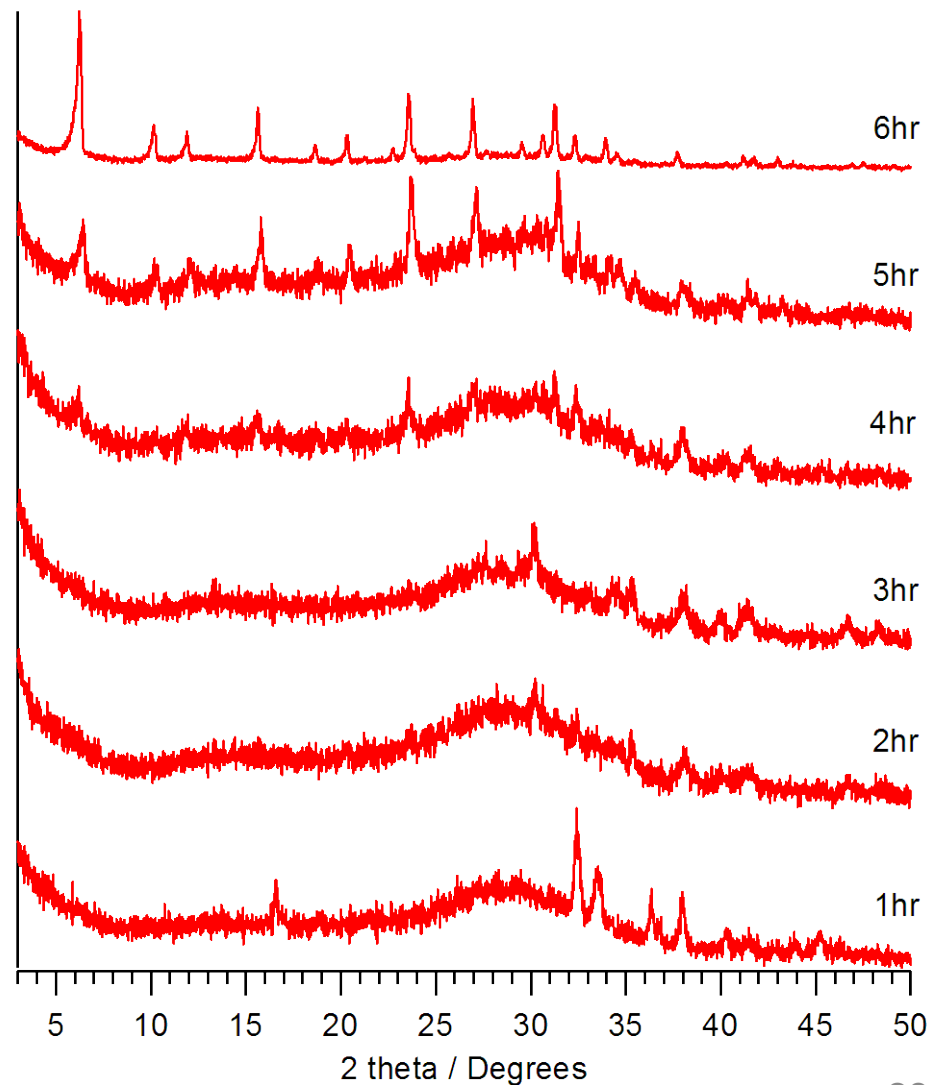
Improve Selectivity Further by Manipulation of Zeolite Layer 27

X-ray Diffraction Patterns indicative of Zeolite Formation

Rapid Method for zeolite growth on polymer (< 1 hour)



Conventional Method requiring 6 hour minimum



Membrane Scale-up

Continuous Membrane Fabrication Machine at OSU



Plans for Future Testing/Development

- **BP2**

- Bench scale membrane synthesis and characterization continue
 - + Membrane performance and stability (with SO₂)
- Continuous membrane fabrication
- Update techno-economic analysis

- **BP3**

- 3 prototype modules for testing with simulated flue gas
- Update and finalize techno-economic analysis
- EH&S evaluation with Gradient Technology and AEP